

DISCLAIMER: The Manganese Consortium believes that the information presented is technically and scientifically correct. However, MARA does not represent or warrant the accuracy of the information contained in this document or its suitability for any general or specific use. The material contained herein is by necessity general in nature; it should not be used or relied upon for any specific or general application without first obtaining competent advice. MARA, its members, staff and consultants specifically disclaim any and all liability or responsibility of any kind for loss, damage, or injury resulting from the use of the information contained in this publication.

#### SAFETY DATA SHEET: MANGANESE SULFATE; MANGANESE (II) SULFATE MONOHYDRATE

Provided in accordance with Article 18(2) of Regulation (EC) No 1272/2008

#### **SECTION 1: IDENTIFICATION**

#### **1.1 Product Identifier:**

Substance name: Manganese (II) Sulfate Monohydrate Other names: Manganese sulphate, MnSO4 EINECS number: 232-089-9 CAS number: 7785-87-7 REACH Registration number: [If applicable] Unique formula identifier (UFI): Not applicable for this substance

#### **1.2** Relevant identified uses of the Substance/Mixture and uses advised against:

Formulation of non-fertiliser products: PC 23 Leather treatment products
Formulation of fertiliser products: PC 12 Fertilisers
Manufacture of fungicides: PC 8 Biocidal products
Production of other manganese-based compounds, intermediate: PC 20
Products such as ph-regulators, flocculants, precipitants, neutralisation agents
Large scale production of penicillin laboratory reagents: PC 29
Pharmaceuticals
Leather tanning: PC 23 Leather treatment products
Surface treatment: PC 9a: Coatings and paints, thinners, paint removal
Printing: PC 18: Ink and toners

Add or delete the above to suit your company's needs.

No known uses advised against

#### 1.3 Details of the supplier of the safety data sheet:(including address, phone numbers etc: Complete as required.

# 1.4 Emergency Telephone: Complete as required (for EU include 112) CIAV # of receiving country.



#### SECTION 2: HAZARD IDENTITIFCATION

#### 2.1 Classification of the substance or mixture:

Classification according to Regulation (EC) No. 1272/2008 [CLP] and the UN GHS: Classified as; STOT RE 2; Eye Dam.1; Aquatic Chronic 2

#### 2.2 Label elements:

Classification	STOT RE 2; Eye Dam.1; Aquatic Chronic 2
<b>Pictogram</b> GHS05, GHS08 and GHS09 respectively	
Signal word	Danger
Hazard statement	<ul> <li>H373: May cause serious damage to the brain through prolonged or repeated exposure via inhalation.</li> <li>H318: Causes serious eye damage.</li> <li>H411: Toxic to aquatic life with long lasting effects</li> </ul>
Precautionary statement Prevention	P260, P273, P280
Precautionary statement Response	P305+351+338
Precautionary Statement Disposal	P501

#### 2.3 Other Hazards:

The substance is an inorganic metallic salt. Based on available information, the substance does not meet the criteria for classification as persistent, bioaccumulative and toxic or very persistent and very bioaccumulative.

Endocrine disrupting properties have not been identified from existing acute or chronic data. Data lacking.

May form explosible dust-air mixture if dispersed.

It is advisable to avoid generating dust as all fine particles have the potential to explode. Dust particles/fumes may damage the eyes. Include other hazards if known.

#### SECTION 3: COMPOSITION/INFORMATION ON INGREDIENTS

#### 3.1 Substance(s) [Amend as appropriate].

Manganese sulfate is an inorganic mono-constituent substance. Its impurities are negligible and do not influence the overall classification.

Chemical name	EC No.	CAS number	Nominal % w/w	REACH
				Registration
				number





Manganese	232-089-9	7785-87-7	>95 - 100% (Complete	XX-XXXXXXX-XX
sulfate			as per your product)	
Water	231-797-2	7732-18-15	<5% (Complete as per	
			your product)	

**3.2 Mixtures:** The substance is not a mixture.

#### SECTION 4: FIRST AID MEASURES

#### 4.1 Description of first aid measures:

#### 4.1.1 General Information

Avoid contact with eyes as the substance can cause serious eye damage. Avoid prolong inhalation. In case of accident exposure or unwellness, seek medical advice immediately.

- **4.1.2 Following Inhalation:** Do not inhale. Wear an appropriate mask. May irritate the lungs and upon prolong exposure may cause subtle neurological effects. Include other relevant information based on your company's procedures.
- **4.1.3 Following Skin Contact:** Wear appropriate protective equipment for good industrial hygiene purposes. Include information based on your company's procedures.
- **4.1.4 Following Eye Contact:** Eye protection is a must. The substance will damage the eyes. Include information based on your company's procedures as well as the recommended specific goggle type.
- **4.1.5** Following Ingestion: Do not ingest. Include other relevant information based on your company's procedures.
- **4.1.6** Self-protection of the first aider: Wear PPE especially eye protection. Include other information based on your company's procedures.
- **4.2 Most important symptoms and effects, both acute and delayed:** Avoid eye contact, as contact with eyes will cause serious eye damage both immediate and delayed.
- **4.3 Indication of any immediate medical attention and special treatment needed:** If in contact with eyes or if accidentally ingested, seek medical attention immediately. Include information based on your company's procedures.

#### SECTION 5: FIRE-FIGHTING MEASURES:

#### 5.1 Extinguishing media:

Include information on an appropriate extinguishing medium and any unsuitable extinguishing media based on your company's procedures.

#### 5.2 Special hazards arising from substance or mixture:

The substance does not decompose naturally. However, combustion produces fumes of metallic oxides and sulfur oxides (Sox). Include any other relevant information.

#### 5.3 Advice for fire-fighters:

Avoid contact with eyes. Flames could produce toxic gases, therefore wear appropriate respiratory protection equipment. Include information based on your company's procedures.



#### SECTION 6: ACCIDENTAL RELEASE MEASURES:

#### 6.1 Personal precautions, protective equipment and emergency procedures:

#### 6.1.1 For non-emergency personnel:

- a) Dust mask and goggles are must. Overalls are encouraged for good industrial hygiene. see section 8 for more details.
- b) Must have dust control and sufficient ventilation. Avoid all ignition sources.
- c) In the event of any accidental release, evacuate the area and consult trained personnel's Amend as per your company procedures.

**6.1.2 For emergency responders:** Remove persons to safety. Isolate hazard area and deny entry. Ventilate closed spaces before entering. Use personal protective equipment, especially goggles and mask, see section 8 – Amend as per your company procedures.

#### 6.2 Environmental precautions:

Substance is considered an environmental hazard. Therefore, keep away from drains/waterways. Water treatment before disposal is essential. Include other information based on your company's procedures.

#### 6.3 Methods and material for containment and cleaning up:

In the event of a spill, collect contaminated material and put in appropriate containers for disposal. Dispose of as special waste in compliance with local and national regulations.

**6.3.1** For containment: Collect in closed and suitable containers for disposal or reuse Include other information based on your company's procedures.

**6.3.2** For cleaning up: Clean contaminated objects and areas thoroughly observing environmental regulations – Amend as per company procedures-Include cleaning and vacuuming techniques.

**6.3.3 Other information:** The substance is hazardous to the environment. Include information based on your company's procedure such as clean-up techniques/materials never to be used.

6.4 **Reference to other sections:** For Personal protective equipment and appropriate disposal: see section 8 and 13.

#### Section 7: HANDLING AND STORAGE:

#### 7.1 Precautions for safe handling:

#### 7.1.1 Recommendations:

- a) Use only in well ventilated areas. Avoid generating dust as dust can easily enter the eyes. Wear personal protective clothing (see Section 8). Include other information based on your company's procedures.
- **b)** Avoid handling with incompatible substances/mixtures: Acids (List incompatible substances if known)
- c) Avoid dust generating operations or must be carried out in properly ventilated areas while wearing appropriate PPE.
- **d)** Capture dust if possible and compress it into pellets to minimize environmental exposure- Amend as per company procedures.

#### 7.1.2 Advice on general occupational hygiene:

- a) Do not eat, drink or smoke in work areas.
- **b)** Wash hands before and after use and keep them dry.
- c) Remove contaminated clothing and personal protective equipment before entering eating areas Include other information based on your company's procedures.



Association

**d)** Capture dust if possible and if generated, vacuum and compress into pellets to minimize environmental exposure- Amend as per company.

#### 7.2 Conditions for safe storage, including any incompatibilities:

#### 7.2.1 Specific storage requirements:

#### a) Risk management associated to physical and chemical properties

i) Explosive atmosphere: The substance is not explosive, however, store away from explosive materials

**ii) Corrosive conditions:** The substance does not corrode metal, hence no adverse corrosive effects are expected

**iii) Flammability hazard:** The substance is not flammable, however, keep away from flammable materials

iv) Incompatible substances or mixtures Acids (List if known): Include information based on company's procedures

v) Evaporative conditions: The substance does not evaporate. Avoid storage around organic evaporative materials/substances.

vi) Potential ignition sources: Keep away from ignition sources

**b)** How to control effects from environmental conditions: (i) Weather conditions, (ii) ambient pressure, (iii) varying temperatures, (iv) sunlight, (v) humidity and (vi) vibration do not affect the integrity of the substance. However, storage environments should be cool – Amend as per your company's procedures.

c) How to maintain the integrity of the substance: The substance is very stable under normal conditions of use. It does not decompose or disintegrate.(i) Stabilisers and (ii) antioxidants are not required

e) Other advise

i) Ventilation requirements: Ensure adequate ventilation and store in a dry and cool place.

**ii) Specific designs for storage:** Keep/store only in original container/packaging. Include other information based on your company's procedures.

**iii)** Quantity limits under storage conditions: There is no limitation as the substance does not pose any physical and chemical hazards.

iv) Packaging compatibility: Store in original/similar packaging. Protect container/packaging against damage – Amend as per company's procedures.

#### 7.3 Specific end uses(s):

Recommendations: Observe instructions for use and see exposure scenarios – Annex 1

#### SECTION 8: EXPOSURE CONTROLS/ PERSONAL PROTECTION:

**8.1 Control Parameters:** 

8.1.1 Occupational exposure limits: The EU SCOEL OEL values for Manganese and its inorganic compounds are 0.2mg/m3 – inhalable and 0.05mg/m3 respirable.

**8.1.1.1** National limits: Include other relevant country specific workplace limits.

**8.1.1.2** Union limits: 0.2mg/m3 inhalable and 0.05mg/m3 respirable

**8.1.1.3** Any other national exposure limit values: Include if available.

**8.1.1.4** Union Biological limit values: No union biological limit values exist for Inorganic manganese and its compounds.

8.1.1.5 Any other national biological limit values: Include if available.



Association

**8.1.2** Monitoring Procedures: Dust monitoring is recommended, provide methodology as per national laws/company procedures.

**8.1.3** Formation of air contaminates: The substance does not produce air contaminants under normal conditions of use. OEL/BLV are therefore not provided Amend as per your company's use.

8.1.4 Derived No Effects Limits (DNELs)/Predicted No Effects Concentrations (PNECs):

#### Hazard Assessment conclusion for Workers: DNELS

Route	Type of effect	Hazard conclusion	Most sensitive endpoint
Inhalation	Systemic effects - Long- term	other toxicological threshold 0.2mg/m <sup>3</sup>	
Inhalation	Systemic effects - Acute	no-threshold effect and/or no dose-response information available	
Inhalation	Local effects - Long-term	other toxicological threshold 0.2mg/m <sup>3</sup>	
Inhalation	Local effects - Acute	no hazard identified	
Dermal	Systemic effects - Long- term	DNEL (Derived No Effect Level) 0.00414mg/kg bw/day	
Dermal	Systemic effects - Acute	no-threshold effect and/or no dose-response information available	
Dermal	Local effects - Long-term	no-threshold effect and/or no dose-response information available	
Dermal	Local effects - Acute	no-threshold effect and/or no dose-response information available	
Eyes	Local effects	no hazard identified	



Manganese REACH Administration Association

#### Hazard Assessment conclusion for the Environment: PNECs

Compartme nt	Hazard conclusion	Remarks/Justification
Freshwater	(freshwater): 0.03mg/L	Assessment factor: 50 Extrapolation method: assessment factor PNEC aqua (freshwater) The lowest chronic NOEC value from the dataset was obtained in the brook trout study (Davies & Brinkman 1998). This study was conducted on MnSO4 and the NOEC was 0.55 mg Mn/L. This is equivalent to 1.51 mg/L of MnSO4 when a molecular weight correction is made. PNEC intermittent release hazard assessment conclusion: PNEC aqua (intermittent releases) PNEC intermittent release assessment factor: 100.0 PNEC intermittent release extrapolation method: assessment factor PNEC intermittent release justification: The lowest acute L(E)C50 value from the dataset was obtained in the rainbow trout (Davies & Brinkman 1998). This study was conducted on MnSO4 and the LC50 was 3.2 mg Mn/L. This is
Marine water	PNEC aqua (marine water): 0.0004mg/L Intermittent releases:	MnSO4 and the LC50 was 3.2 mg Mn/L. This is equivalent to 8.79 mg/L of MnSO4 when a molecular weight correction is made. Assessment factor: 50 Extrapolation method: assessment factor PNEC aqua (marine water)



		Based on the freshwater HC5 and the long-term NOEC from a study on the marine species; Pacific oyster (NOEC of 0.02 mg Mn/L).
Sediments (freshwater)	PNEC sediment (freshwater): 0.0114mg/kg sediment dw	Assessment factor: 50 Extrapolation method: assessment factor PNEC sediment (freshwater) Based on the lowest endpoint (NOEC of 0.57 mg Mn/kg sediment dwt) from studies on two sediment dwelling organisms
Sediments (marine water)	PNEC sediment (marine water): 0.00114mg/kg sediment dw	Assessment factor: 500 Extrapolation method: equilibrium partitioning method PNEC sediment (marine water) Based on the freshwater endpoints with an increased AF factor.
Sewage treatment plant	PNEC STP: 56mg/L	Assessment factor: 10 Extrapolation method: assessment factor PNEC STP Activated sludge respiration/inhibition test; NOEC = 560 mg MnSO4/L



Soil	PNEC soil: 25.1mg/kg soil dw	Assessment factor: 10 Extrapolation method: assessment factor PNEC soil
		Based on the lowest NOEC (251 mg Mn/kg soil d.w.) from a range of long-term studies
Air	no hazard identified:	
Secondary poisoning	no potential for bioaccumulation:	Bioaccumulation of Mn is not expected to occur. Hence no secondary poisoning risk exists.

**8.1.5** Control banding: A control banding approach is not used to decrease risk management measure during the use of this substance for the uses specified in section 1.2.

#### 8.2 Exposure controls: See Exposure scenarios on, Annex 1

**8.2.1** Appropriate engineering controls: Dust is trapped and recycled where possible. Wastewater is collected for treatment and recycled. Amend as per your company's procedures.

**8.2.2** Individual protective measures: Overalls, goggles and masks are mandatory during use.

**8.2.2.1** Other non-personal protection: Good industrial hygiene is a must. Keep and use in well ventilated areas. See section 5 for more information Amend as per your company's procedures.

**8.2.2.2** CEN stand requirement for protective equipment: Goggles and masks are a must. (Please state the quality/standard/thickness of the personal protective equipment used by your organisation)

a) Eye/face protection: Goggles and masks are a must. complete as per your company procedures e.g type of goggles



**b)** Skin protection: Overalls, gloves and boots are not mandatory; however, they are encouraged for good industrial hygiene (Please specify type of overall, gloves, boots including the thickness of material)

c) Overalls, gloves and boots are not mandatory. However, they are encouraged for good industrial hygiene. (Please specify type of overall, gloves, boots including the thickness of material)

d) Respiratory protection: N95 Mask (Amend as per your company procedures)e) Thermal hazards: Not applicable

**8.2.3** Environmental exposure controls: The substance is toxic to the environment according to EU harmonised classification. See Annex I, Exposure scenarios (Please include environmental controls employed by your company)

# SECTION 9: PHYSICAL AND CHEMICAL PROPERTIES: [the information below is based on available literature and studies]

9.1 General information	
State	Solid
Colour	Light Pink
Odour	Odourless
Melting point (Mpt) / Freezing point	>723 K (>450°C), Regulation (EC) No. 440/2008,
	Annex, A1
Boiling point or initial boiling point/boiling	Melting pt >300°C, hence study not applicable
range	
Flammability of solids	Not flammable, Regulation (EC) No. 440/2008,
	Method A10
Lower and upper explosion limit	Not applicable to solids
Flash Point	Not applicable to solids
Auto-ignition temperature	Not applicable to solids
Decomposition temperature	Not applicable to inorganic solids
рН	Include if known (where the substance is a solid,
	the pH of an aqueous solution at a given
	concentration shall be indicated)
Kinematic viscosity	Not applicable to solids
Water Solubility	Soluble: 42.5 to 45.0% at 20±0.5°C., Regulation
	(EC) No. 440/2008, Annex A6
Partition Coefficient	Not applicable for inorganic substances
Vapour pressure	Study not conducted as Mpt >300°C
Density/Relative density	2.93 at 220. +/- 0.5°C, Regulation (EC) No.
	440/2008, Annex, A3
Particle characteristics	Data lacking
9.2 Other information	None
9.2.1 Physical hazard classes	
-	
Explosive properties	Predicted to be non-explosive



Flammable gases	Not applicable as the substance is a solid
Aerosols	Not applicable under normal conditions of use
Oxidizing gases	Not applicable as the substance is a solid
Gases under pressure	Not applicable as the substance is a solid
Flammability of liquids/Solids	Not flammable
Self-reactive substances and mixtures	Not self-reactive
Pyrophoric liquids	Not applicable as the substance is a solid
Pyrophoric solids	Does not have pyrophoric properties
Self-heating substances and mixtures	Spontaneous ignition does not occur
Substances and mixtures which emit flammable	Predicted not to emit flammable gases upon
gases in contact with water	contact with water
Oxidising Liquids/solids	Non oxidising, Method A17
Organic peroxides	Not applicable to inorganic substances
Corrosive to metals	The substance is not corrosive to metals
Desensitised explosives	Not applicable
9.2 Other information	No additional information relevant to the safe
	use of the substance

SECTION 10: STABILITY AND REACTIVITY: [Amend information below to conform to your company information]

**10.1 Reactivity:** No specific test data related to reactivity available for this substance.

10.1.1 Reactivity Hazard of substance: Not applicable for inorganic substances

10.1.2 **Reactivity hazard of mixture:** Not applicable as the substance is not a mixture.

**10.2** Chemical stability: The substance is chemically stable under recommended conditions of storage, use and temperature.

**10.3 Possibility of Hazardous reaction:** No hazardous reaction when handled and stored according to provisions.

- **10.4** Conditions to avoid: Include your company's information.
- **10.5** Incompatible Materials: Acids. Include your company's information.
  - **10.6 Hazardous decomposition products:** Does not decompose when used for intended uses. Include your company's information.

# SECTION 11: TOXICOLOGICAL INFORMATION: [The information in this section is from experimental data and other available literature]

#### **11.1** Information on toxicological effects:

#### a) Acute toxicity:

Acute oral toxicity: Not harmful if ingested (LD50 >2000mg/kgbw) Acute dermal toxicity: Not a dermal irritant

Acute inhalation toxicity: No adverse effects observed.

#### b) Skin corrosion/irritation:

Not corrosive to the skin (Skinethic reconstituted human epidermal model) c)Serious eye damage/irritation:



Substance causes serious eye damage (study according to OECD guideline 405 and EU method B. 5, GLP)

#### d) Respiratory or skin sensitization:

Not a skin sensitizer (available literature similar to OECD guideline 429 and EU method B.42, Local lymph node assay). There is no information available for respiratory sensitization. However, it is predicted not to be a respiratory sensitizer.

#### e) Germ cell mutagenicity:

Data lacking for the substance as such due to its corrosivity nature. However, data on MnCl2 – a very soluble salt considered as a worse-case evaluation concludes – No concerns on mutagenicity based on studies below:

- Ames test with S. typhimurium TA 98, TA 100, TA 1535, TA 1537, E coli WP2 uvrA (Met. act.: with and without) (OECD TG 471, EU method B13 and GLP); No toxicity was observed up a concentration of 5000 ug/plate.
- Mammalian cell gene mutation assay with mouse lymphoma L5178Y cells (met. act.: with and without) (OECD 476 and GLP); Negative for mouse lymphoma Cytoxicity: Yes, induced toxicity was not at the highest dose.
- In-vitro mammalian chromosome aberration test with human lymphocytes (Met. act.: with and without) (OECD guideline 473 and GLP). Negative for lymphocytes. Cytotoxicity: Yes

#### f) Carcinogenicity:

Carcinogenicity report (NTP, 1993) on the substance as such and an expert review by Jenkinson, 2009 on genotoxicity as well as peer review article (Assem et al, 2011) on several inorganic manganese-based substances concluded – no concerns, carcinogenicity in humans is not expected.

#### g) Reproductive toxicity:

Two Generation reprotoxicity study on the male/female rats using MnCL2 via inhalation (OECD guideline 416, GLP): concluded: No treatment related effects at 20 mg/m<sup>3</sup> air in F0, F1 and F2 generations (Jardine L, 2013 and McGough & Jardine, 2017) - Not toxic to reproduction

A Prenatal-developmental toxicity study using MnCl2 via inhalation (OECD 414, GLP): concluded no fetal abnormalities at not specified at 15 mg/m<sup>3</sup> (Dettwiler M, 2016)

#### h) Specific target organ toxicity (Single exposure):

Based on available data the classification criteria are not met.

#### i)Specific target organ toxicity (repeated exposure):

Based on published literature the substance is relatively non-toxic via the oral route however, inhalation exposure is expected to cause harm upon repeated exposure over long periods of time. The substance has a harmonized classification as STOT RE 2.

#### j) Aspiration hazards:

Data lacking

#### **11.2** Information on other hazard

**11.2.1 Endocrine disrupting properties:** The substance is not considered an endocrine disruptor based on available literature – Data lacking.



#### SECTION 12: ECOLOGICAL INFORMATION:

**12.1 Toxicity:** Toxic to aquatic life with long-lasting effects by harmonise classification, which is not applicable based on available data

#### Acute (short-term)

#### Aquatic vertebrates:

- a) Fish: LC50 (96h): 49.9 mg/L Mn based on mortality (test material-MnSO4)
- b) Fish: LC50 (96h): 27.5 mg/L Mn based on: mortality (150 Hardness. 95 % 23.4 31.6 mg/L.) and LL50 (96h): 5.12 mg/L Mn based on: mortality (30 Hardness. 95 % CL 4.6 5.7 mg//L.) (test material -MnCl2 based on readacross)

#### Aquatic invertebrates:

 a) Daphnia magna: LC50 (48h): 9.8 mg/L dissolved (meas. (arithm. mean)) based on: as Mn2+ (Without Food) (test material -MnCl2 based on readacross)

Chronic (long-term) toxicity:

#### Aquatic vertebrates:

- a) Fish early-life stage toxicity: NOEC (4mo): 0.6 mg/L Mn (test material MnCl2 based on readacross)
- **b)** Fish early-life stage toxicity: NOEC (4mo): 2.03 mg/L Mn (test material MnCl2 based on readacross)
- c) Oncorhynchus mykiss (previous name: Salmo gairdneri): Early life: NOEC (4mo): 0.6 mg/L Mn (test material-MnSO4

#### Aquatic invertebrates:

- a) Fresh water Ceriodaphnia dubia: LC50 (48h): 5.7 mg Mn/L (Average) test mat. (estimated) based on soft water survival (test material -MnCl2 based on readacross)
- b) Aquatic worm Aeolosoma sp. : LC50 (48h): 39.46 mg/L dissolved (meas. (arithm. mean)) based on: mortality (95 % CL); LOEC (48h): 53.67 mg/L dissolved (meas. (arithm. mean)) based on: Survival; NOEC (48h): 27.2 mg/L dissolved (meas. (arithm. mean)) based on: Survival (test material MnCl2 based on readacross)
- c) ASRI: No effects were seen on microbial activity (OECD Guideline 209, Activated Sludge, Respiration Inhibition- ASRI) at 3 hours exposure: 560 mg/L test material and EC50 (3h) >1000 mg/L test material (nominal) based on: inhibition of total respiration - respiration rate (test material -MnCl2 based on readacross)

#### **Other Environmental studies**

a) Soil macro-organisms at levels up to 157 mg/kg soil dw Mn during 28 days exposure (Kuperman RG, et al DJ 2002)

Therefore, based on available studies conducted at different trophic levels, the substance is not harmful to aquatic life. However, harmonise classification takes precedence.

12.2 Persistence and degradation	No potential for	According to the Guidance on information requirements and chemical safety assessment,
	persistence	Chapter R.11: PBT assessment, "the PBT and vPVB criteria of Annex XIII to the regulation
		do not apply to inorganic substances". Therefore MnSO4 is not considered to require any
		further assessment of PBT properties.



	bioaccumulation	According to the Guidance on information requirements and chemical safety assessment, Chapter R.11: PBT assessment, "the PBT and vPVB criteria of Annex XIII to the regulation do not apply to inorganic substances". Therefore MnSO4 is not considered to require any further assessment of PBT properties.
,	No potential to move into ground water	Data lacking

#### 12.5 Results of PBT and vPvB assessment:

According to the Guidance on information requirements and chemical safety assessment, Chapter R.11: PBT assessment, "the PBT and vPVB criteria of Annex XIII to the regulation do not apply to inorganic substances". Therefore, MnSO4 is not considered to require any further assessment of PBT properties.

#### 12.6 Endocrine disrupting properties:

The substance is not considered an endocrine disruptor based on available literature – Data lacking.

12.7 Other adverse effects: None known

#### **SECTION 13: DISPOSAL CONSIDERATIONS:** Include your company's information.

**13.1 Waste treatment methods:** Waste disposal in accordance with local and national laws covering waste and dangerous waste. Include additional company specific information.

a) Waste treatment-relevant information: Include your company's /national law information.

b) Physical/chemical properties that affect waste treatment option: None known.

c) Sewage disposal-relevant information: Include your company's/national laws information.

d) Precautions for recommended waste treatment options: Include your company's information.

#### SECTION 14: TRANSPORT INFORMATION:

Transport may take place according to national regulations or land transport (ADR/RID), sea transport (IMDG) or Air transport (ICAO-TI/IATA-DGR).

**14.1 UN Number:** Include number or use not applicable if this is the case.

14.2 UN proper shipping name: Include name or use not applicable if this is the case.

#### 14.3 Transport hazard class: Not hazardous

**14.4 Packaging group:** Include packaging group or use not applicable if this is the case.

14.5 Environmental hazard: Not hazardous to the environment

**14.6 Special precautions for users:** Always transport in close containers, avoid generating dust [Amend as appropriate]



**14.7 Maritime transport in bulk according to IMO instruments** Complete as appropriate or used not applicable if this is the case.

# SECTION 15: REGULATORY INFORMATION: [Delete as appropriate and include regulatory information specific to your country...]

#### 15.1 Safety, health and environmental regulations/legislation for the substance:

**UN GHS - UN Globally Harmonized System of Classification and Labeling of Chemicals (GHS):** According to Chapter 1.5.2 of the UN Globally Harmonized System of Classification and Labeling of Chemicals (GHS) safety data sheets (SDS) are only required for substances and mixtures that meet the harmonized criteria for physical, health or environmental hazards. This substance meets these criteria; hence a safety data sheet is required.

**EU CLP – Classification Labeling and Packaging Regulation:** According to Article 59(2)(b) of (EC) No 1272/2008 (CLP), which amends REACH article 31(1), safety data sheets (SDS) are only required for substances and mixtures/special preparations that meet the harmonized criteria for physical, health or environmental hazards. MnSO4 meets this criterion, hence a SDS according to 453/2010/EC is needed – this template is designed to meet this criteria.

**EU REACH – Registration, Evaluation and Authorisation of Chemicals:** REACH article 31(7) requires relevant exposure scenarios from the Chemical Safety Report (CSR) to be annexed to the SDS. These exposure scenarios are only required for hazard-classified substances or mixtures. This substance is hazard-classified according to CLP; therefore exposure scenarios are required. Ask your REACH/Chemical regulatory team.

**15.2 Chemical Safety Report (CSR):** A chemical safety assessment has been carried for this substance.

#### SECTION 16: OTHER INFORMATION:

- **16** If using this template to develop your company's SDS in the case of a revised safety data sheet, a clear indication of where changes have been made to the previous version of the safety data sheet is required in this section, unless such indication is given elsewhere in the safety data sheet, with an explanation of the changes, if appropriate. A supplier of a substance or mixture shall be able to provide an explanation of the changes upon request.
- 17 A key/legend to abbreviations and acronyms used in the SDS should be added in this section.18 Key Literature:
  - 1. Assem, F. L., et al, (2011); The Mutagenicity and carcinogenicity of inorganic manganese compounds: A synthesis of the evidence, Journal of toxicology and environment, part B
  - 2. Adhikari S, Naqvi AA, Pani KC, Pillai BR, Jena JK & Sarangi N (2007). Effect of Manganese and Iron on Growth and Feeding of Juvenile Giant River Prawn, Macrobachium rosenbergii (De-Man). Journal of the World Aquaculture Society. 38(1):161-168.
  - 3. Akoume MY, Perwaiz S, Yousef IM and Plaa GL (2003). Synergistic role of 3-hydroxy-3methylglutaryl coenzyme A reductase and cholesterol 7alpha-hydroxylase in the pathogenesis of manganese-bilirubin induced cholestasis in rats. Toxicol Sci 73:378-385.
  - 4. Andersen O (1983). Effects of coal combustion products and metal compounds on sister chromatid exchange (SCE) in a macrophage-like cell line. Environmental Health Perspectives. 47:239-253.



5. Aschner M, Erikson KM and Dorman DC (2005). Manganese dosimetry: species differences and implications for neurotoxicity. Crit Rev Toxicol 35:1-32.

6. Atwal SS and Tremain SP (2009). MnSO4: Determination of Oxidising Properties (Solids). Testing laboratory: Harlan Laboratories Limited, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0018. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2009-09-14.

- Atwal SS and Woolley SM (2009). MnSO4: Determination of Melting/Freezing Temperature and Flammability (Solids). Testing laboratory: Harlan Laboratories Limited, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0041. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2009-08-19.
- 8. Bales CW, Freeland-Graves JH, Lin PH, Stone JM and Dougherty V (1987). Plasma Uptake of Manganese Influence of Dietary factors. American Chemical Society, Washington DC.
- 9. Baranowska H, Ejchart A and Putrament A (1977). Manganese mutagenesis in yeast. V. on mutation conversion induction in nuclear DNA. Mutation Research. 42: 343-348.
- Basketter A. D, Lea L. J, Cooper K. J, Ryan C. A, Gerberick G. F, Dearman R. J, Kimber I (1999). Identification of metal allergens in the local lymph node assay. American journal of contact dermatitis, 10:207-212.
- 11. Bataineh H, Al-Hamood MH and Elbetieha AM (1998). Assessment of aggression, sexual behaviour and fertility in adult male rat following long-term ingestion of four industrial metal salts. Human and Experimental Toxicology, 17:570-576.
- 12. Bataineh HN, Bateineh ZM & Daradka H (2007). Short-term exposure of female rats to industrial metal salts: effect on implantation and pregnancy. Reproductive Medicine and Biology, 6: 179-183.
- 13. Biesinger KE & Christensen GM (1972). Effects of Various Metals on Survival, Growth, Reproduction, and Metabolism of Daphnia magna. Journal of the Fisheries Research Board of Canada.29:1691-1700.
- 14. Bouchard M, Laforest F, Vandelac L, Bellinger D & Mergler D. (2007). Hair Manganese and Hyperactive Behaviors: pilot Study of School-Age Children Exposed through Tap Water. Environ. Health Perspec., 115(1):122-7.
- Cardwell A (2009a). Evaluation of Acute Manganese Toxicity to the Aquatic Oligochaete, Aeolosoma sp. testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas Street SW, Albany, or 97321. Report no.: 598-5231-002. Owner company: International Manganese Institute,17, rue Duphot, 75001 Paris, France. Report date: 2009-03-06.
- Cardwell A (2009b). Evaluation of Chronic Toxicity of Manganese to the Aquatic Oligochaete, Aeolosoma sp. Testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas Street SW, Albany, OR 97321. Report no.: 598-5231-002. Owner Company:



International Manganese Institute, 17 rue Duphot, 75001 Paris, France. Report date: 2009-03-06.

17. Cardwell A (2009c). 42- day Chronic Toxicity of Manganese to the freshwater amphipod crustacean, Hyalella azteca. Testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas St. SW, Albany, Oregon 97321, USA. Report no.: 598-5232-002. Owner company: International Manganese Institute, 17, rue Duphot, 75001 Paris, Franc. Report date: 2009-03-06.

- Cardwell A (2009d). Life-Cycle Toxicity of Manganese to the Midge, Chironomus tentans. Testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas St. SW, Albany, Oregon 97321, USA. Report no.: 598-5231-002. Owner Company: International Manganese Institute, 17 rue Duphot, 75001 Paris, France. Report date: 2009-03-06.
- 19. Chandra SV, Saxena DK and Hasan MZ (1975). Effect of zinc on manganese induced testicular injury in rats. Ind. Health, 13: 51-56.
- 20. Chen MK, Lee J, McGlothan J, Furukawa E, Adams R, Alexander M, Wong, D & Guilarte T. (2006). Acute manganese administration alters dopamine transporter levels in the non-human primate striatum. NeuroToxicity, 27(2): 229-236.
- 21. Davies H & Brinkman S (1995). Acute toxicity of manganese to brown trout (salmo trutta) in hard water. Federal aid Project #F-243, Colorado Division of Wildlife, Fort Collins, Colorado.
- 22. Davies P & Brinkman S (1994). Acute and chronic toxicity of manganese to exposed and unexposed Rainbow and brown trout. Federal aid project F-243-1, Colorado Division of Wildlife, Fort Collins, Colorado.
- 23. Davies P and Brinkman S (1994). Acute and chronic toxicity of manganese to exposed and unexposed Rainbow and brown trout. Federal aid project F-243-1, Colorado Division of Wildlife, Fort Collins, Colorado.
- 24. Davies PH (1980). Investigations on the toxicity of metals to fish. Colorado division of wildlife, Fish research section, Colorado.
- 25. Davies PH, Brinkman, S & McIntyre M (1998). Toxicity of Manganese to Early life stage and Fry of Brook Trout (Salvenlinus fontinalis) and Rainbow trout (Oncorhynchus mykiss) in water hardnesses of 30 and 150 mg/L. Federal aid project #F-243R-5, Colorado Division of Wildlife, Fort Collins, Colorado.
- 26. Davies PH, Brinkman, S and McIntyre M (1998). Toxicity of Manganese to Early life stage and Fry of Brook Trout (Salvenlinus fontinalis) and Rainbow trout (Oncorhynchus mykiss) in water hardnesses of 30 and 150 mg/L. Federal aid project #F-243R-5, Colorado Division of Wildlife, Fort Collins, Colorado.
- 27. Davis H, Brinkman S and McIntyre M (1998). Toxicity of manganese and zinc to Boreal toad tadpoles (Bufo boreas). Federal aid project #F-243R-5, Colorado Division of Wildlife, Fort Collins, Colorado.
- 28. De Meo M, Laget M, Castegnaro M & Dumenil G (1991). Genotoxic activity of potassium permanganate in acidic solutions. Mutation Research. 260: 295-306.



29. De Meo M, Laget M, Castegnaro M and Dumenil G (1991). Genotoxic activity of potassium permanganate in acidic solutions. Mutation Research. 260: 295-306.

30. Del Carmen Hernandez Soriano M, Martens J & Smolders E (2009). Toxicity of Manganese to micro-organism and to higher plants in soils. Testing laboratory: K. U. Leuven, Department of Earth and Environmental Sciences, Division of Soil and Water Management, Kasteelpark Arenberg 20, 3001 Heverlee, Belgium. Owner Company: Manganese Reach Administration, 17 rue Duphot Paris, France 75001. Report date: 2009-10-01.

- 31. del Carmen Hernandez Soriano M, Mertens J & Smolders E (2009a). Toxicity of Manganese to micro-organisms and to higher plants in soil. Testing laboratory: K. U. Leuven, Department of Earth and Environmental Sciences, Division of soil and Water management, Kasteelpark Arenberg 20, 3001 Heverlee, Belgium. Owner Company: Manganese Reach Administration Association, 17 rue Duphot Paris, France 75001. Report date: 2009-10-01.
- 32. del Carmen Hernandez Soriano M, Mertens J & Smolders E (2009b). Toxicity of Manganese to micro-organisms and to higher plants in soils. Testing laboratory: K. U. Leuven, Department of Earth and Environmental Sciences, Division of Soil and Water Management, Kasteelpark Arenberg 20, 3001 Heverlee, Belgium. Owner Company: Manganese Reach Administration Association, 17 rue Duphot Paris, France 75001. Report date: 2009-10-01.
- 33. Dobson A, Weber S, Dorman D, Lash L, Erikson K & Aschner M (2003). Oxidative Stress Is Induced in the Rat Brain Following Repeated Inhalation Exposure to Manganese Sulfate. Biological Trace Element Research, 93: 113-125.
- 34. Dorman DC, McElveen AM, Marshall MW, Parkinson CU, Arden James R, Struve MF and Wong BA (2005). Maternal-fetal distribution of manganese in the rat following inhalation exposure to manganese sulfate. Neurotoxicology 26:625-632.
- 35. Dorman DC, McElveen AM, Marshall MW, Parkinson CU, James RA, Struve MF & Wong BA (2005b). Maternal-fetal Distribution of Manganese in the Rat Following Inhalation Exposure to Manganese Sulfate. NeuroToxicology, 26: 625-632.
- Dorman DC, McElveen AM, Marshall MW, Parkinson CU, James RA, Struve MF and Wong BA (2005a). Tissue manganese concentrations in lactating rats and their offspring following combined in utero and lactation exposure to inhaled manganese sulfate. Toxicol Sci 84:12-21.
- 37. Dorman DC, McManus BE, Marshall MW, James RA and Struve MF (2004). Old age and gender influence the pharmacokinetics of inhaled manganese sulfate and manganese phosphate in rats. Toxicol Appl Pharmacol 197:113-124.
- 38. Dorman DC, McManus BE, Parkinson CU, Manuel CA, McElveen AM and Everitt JI (2004). Nasal toxicity of manganese sulfate and manganese phosphate in young male rats following subchronic (13-week) inhalation exposure. Inhal Toxicol 16:481-488.
- 39. Dorman DC, Struve MF and Wong BA (2001). Pharmacokinetic Factors That Influence Manganese Delivery to the Brain. CIIT Activities 21:1-11.



40. Dorman DC, Struve MF, Clewell HJ, 3rd and Andersen ME (2006). Application of pharmacokinetic data to the risk assessment of inhaled manganese. Neurotoxicology 27:752-764.

- 41. Dorman DC, Struve MF, James RA, Marshall MW, Parkinson CU and Wong BA (2001). Influence of particle solubility on the delivery of inhaled manganese to the rat brain: manganese sulfate and manganese tetroxide pharmacokinetics following repeated (14-day) exposure. Toxicol Appl Pharmacol 170:79-87.
- 42. Dorman DC, Struve MF, James RA, McManus BE, Marshall MW and Wong BA (2001). Influence of dietary manganese on the pharmacokinetics of inhaled manganese sulfate in male CD rats. Toxicol Sci 60:242-251.
- 43. Dorman DC, Struve MF, Marshall MW, Parkinson CU, Arden James R and Wong BA (2006). Tissue manganese concentrations in young male Rhesus monkeys following subchronic manganese sulphate inhalation. Toxicological Sciences. 92(1); 201-210. Testing laboratory: CIIT Centers for Health Research, 6 Davis Drive, PO BOX 12137, Research Triangle Park, North Carolina 27709-2137.
- 44. Dorman DC, Struve MF, Marshall MW, Parkinson CU, James RA and Wong BA (2006). Tissue manganese concentrations in young male rhesus monkeys following subchronic manganese sulfate inhalation. Toxicol Sci 92:201-210.
- 45. Dorman DC, Struve MF, Wong BA, Dye JA and Robertson ID (2006). Correlation of brain magnetic resonance imaging changes with pallidal manganese concentrations in rhesus monkeys following subchronic manganese inhalation. Toxicol Sci 92:219-227.
- 46. Erikson KM, Dorman DC, Lash L & Aschner M (2007). Manganese Inhalation by Rhesus Monkeys is Associated with Brain Regional Changes in Biomarkers of Neurotoxicity. Toxicological Sciences, 97(2): 459-466.
- 47. Erikson KM, Dorman DC, Lash LH, Dobson AW & Aschner M (2004). Airborne Manganese Exposure Differentially Affects End Points of Oxidative Stress in an Age- and Sex- Dependent Manner. Biological Trace Element Research, 100: 49-62.
- 48. Fechter LD (1999). Distribution of manganese in development. Neurotoxicology 20:197-201.
- 49. Finley JW (1999). Manganese absorption and retention by young women is associated with serum ferritin concentration. Am J Clin Nutr 70:37-43.
- 50. Finley JW, Penland JG, Pettit RE and Davis CD (2003). Dietary manganese intake and type of lipid do not affect clinical or neuropsychological measures in healthy young women. J Nutr 133:2849-2856.
- Flanders L (2009). MnCl2 (Eramet): L5178Y TK +/- Mouse Lymphoma Assay. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0037. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2009-11-17.
- 52. Galloway SM, Armstrong MJ, Reuben C, Colman S, Brown B, Cannon C, Bloom AD, Nahamura F, Ahmed M, Duk S, Rimpo J, Margolin BH, Resnick MA, Anderson B and Zeiger E (1987).



Chromosome Aberrations and Sister Chromatid Exchanges in Chinese Hamster Ovary Cells: Evaluations of 108 Chemicals. Environmental and Molecular Mutagenesis, 10(10): 1-175.

53. Graham R (2009). Early Life-Stage Toxicity of Manganese to the Zebrafish (Danio rerio) Under Flow-Through Conditions. Testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas St. SW, Albany, Oregon 97321-9487, USA. Report no.: 598-5231-002. Owner Company: International Manganese Institute, 17, rue Duphot, 75001 Paris, France. Report date: 2009-03-01.

- Griffiths DR (2010). MnSO4: ACUTE INHALATION TOXICITY (NOSE ONLY) STUDY IN THE RAT. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire DE72 2GD, U. K. Report no.: 2702/0139. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2010-04-27.
- 55. Hernandez Soriano MDC, Degryse F and Smolders E (2010). Adsorption and desorption screening of Mn in soils (Draft report). Testing laboratory: K. U. Leuven, Department of Earth and Environmental Sciences Division of Soil and Water Management Kasteelpark Arenberg 20 3001 Heverlee Belgium.
- 56. Ikarashi Y, Tsuchiya T and Nakamura A (1992). Detection of contact sensitivity of metal salts using the murine local lymph node assay. Toxicology letters, 62: 53-61.
- 57. Ji F, Luo XG, Lu L, Liu B and Yu SX (2006). Effect of manganese source on manganese absorption by the intestine of broilers. Poult Sci 85:1947-1952.
- 58. Joardar M and Sharma A (1990). Comparison of the clastogenicity of inorganic Mn administered in cationic and anionic forms in vivo. Mutation Research. 240: 159-163.
- 59. Järvinen R & Ahlström A (1975). Effect of dietary manganese level on tissue manganese, iron, copper and zinc concentrations in female rats and their fetuses. Medical Biology 53: 93-99.
- 60. Kuperman RG, Checkai RT, Phillips CT, Simini M, Speicher JA and Barclift DJ (2002). Toxicity Assessments of Antimony, Barium, Beryllium, and Manganese for Development of Ecological Soil Screening Levels (Eco-SSL) Using Enchytraeid Reproduction Benchmark Values. Edgewood chemical biological center, US army soldier and biological chemical command. Report no.: ECBC-TR-324.
- 61. Lasier PJ, Winger PV & Bogenrieder KJ (2000a). Toxicity of Manganese to Ceriodapnia dubia and Hyalella azteca. Arch. Environ. Contam. Toxicol.38:298-304.
- 62. Lasier PJ, Winger PV & Bogenrieder KJ (2000b). Toxicity of Manganese to Ceriodaphnia dubia and Hyalella azteca. Arch. Environ. Contam. Toxicol.38:298-304.
- 63. Lasier PJ, Winger PV & Bogenrieder KJ (2000c). Toxicity of Manganese to Ceriodaphnia dubia and Hyalella azteca. Arch. Environ. Contam. Toxicol.38:298-304.
- 64. Lauwerys R, Roels H, Genet P, Toussaint G, Bouckaert A & de Cooman S (1985). Fertility of male workers exposed to mercury vapor or to manganese dust: A questionnaire study. American Journal of Industrial Medicine, 7:171-176.



65. Lewis M (1978). Acute toxicity of copper, zinc and manganese in single and mixed salt solutions to juvenile longfin dace, Agosoa chrysogaster. J. Fish Biolo.13:695-700.

66. Marzin DR and Phi HV (1985). Study of the mutagenicity of metal derivatives with Salmonella typhimurium TA 102. Mutation Research. 155: 49-51.

- 67. Matrone G, Hartman RH and Clawson AJ (1959). Studies of a manganese-iron antagonism in the nutrition of rabbits and baby pigs. J Nutr 67:309-317.
- 68. Maynard LS and Cotzias GC (1955). The partition of manganese among organs and intracellular organelles of the rat. J Biol Chem 214:489-495.
- 69. McGough D, Jardine L. A two-generation inhalation reproductive toxicity study upon the exposure to manganese chloride. Neurotoxicology. 2017 Jan;58:194-202. doi: 10.1016/j.neuro.2016.09.017. Epub 2016 Sep 23.
- 70. Menezes LM, Campos LC, Quintao CC and Bolognese AM (2004). Hypersensitivity to metals in orthodontics. American Journal of Orthodontics and Dentofacial Orthopedics. 126:58-64.
- 71. Miller KB, Caton JS, Schafer DM, Smith DJ and FInley JW (2000). High dietary manganese lowers heart magnesium in pigs fed a low-magnesium diet. Journal of Nutrition, 130:2032-2035.
- Morera D (2009). Toxicity of Manganese to Pseudokirchneriella subcapitata under static test conditions. Testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas street SW, Albany, OR 97321- 9487, USA. Report no.: 598-5231-002. Owner Company: International Manganese Institute, 17 rue Duphot, 75001 Paris, France. Report date: 2009-03-01.
- 73. Morris A & Durward R (2009). MnCl2 (Eramet): Chromosome Aberration Test in Human Lymphocytes In Vitro. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0036. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2009-11-23.
- 74. Mortelmans K, Haworth S, Lawlor T, Speck W, Tainer B and Zeiger E (1986). Salmonella mutagenicity tests: II. Results from the testing of 270 chemicals. Environmental Mutagenesis 8(7): 1-119.
- 75. NTP (1993a). NPT TR 428: NTP technical report on the toxicology and carcinogenesis of manganese (II) sulphate monohydrate in F344/N rats and B6C3F1 mice (feed studies). National Toxicology Program, P. O Box 12233, Research Triangle Park, NC 27709. Testing laboratory: Studies conducted on behalf of the National Toxicology Program by: Gulf South Research Institute, Battelle Columbus Laboratories, Experimental Pathology Laboratories, Inc., Integrated Laboratory Systems, Biotechnical Services, Inc. Report no.: NIH publication number: 94-3159.
- 76. NTP (1993b). NPT TR 428: NTP technical report on the toxicology and carcinogenesis of manganese (II) sulphate monohydrate in F344/N rats and B6C3F1 mice (feed studies). Testing laboratory: Studies conducted on behalf of the National Toxicology Program by: Gulf South Research Institute, Battelle Columbus Laboratories, Experimental Pathology



Laboratories, Inc., Integrated Laboratory Systems, Biotechnical Services, Inc. Report no.: NIH publication number: 94-3159.

77. NTP (1993c). NPT TR 428: NTP technical report on the toxicology and carcinogenesis of manganese (II) sulphate monohydrate in F344/N rats and B6C3F1 mice (feed studies). Testing laboratory: Studies conducted on behalf of the National Toxicology Program by: Gulf South Research Institute, Battelle Columbus Laboratories, Experimental Pathology Laboratories, Inc., Integrated Laboratory Systems, Biotechnical Services, Inc. Report no.: NIH publication number: 94-3159.

- 78. Nishioka H (1975). Mutagenic activities of metal compounds in bacteria. Mutation Research 31:185-189.
- 79. Normandin L, Ann Beaupre L, Salehi F, St -Pierre A, Kennedy G, Mergler D, Butterworth RF, Philippe S and Zayed J (2004). Manganese distribution in the brain and neurobehavioral changes following inhalation exposure of rats to three chemical forms of manganese. NeuroToxicology 25:433-441.
- 80. Olivier P & Marzin D (1987). Study of the genotoxic potential of 48 inorganic derivatives with the SOS chromatest. Mutation Research. 189: 263-269.
- 81. Papavasiliou PS, Miller ST and Cotzias GC (1966). Role of liver in regulating distribution and excretion of manganese. Am J Physiol 211:211-216.
- 82. Parry JM (1977). The use of yeast cultures for the detection of environmental mutagens using a fluctuation test. Mutation Research. 46:165-175.
- 83. Phillips CT, Checkai RT, Kuperman RG, Simini M, Speicher JA and Barclift DJ (2002). Toxicity assessments of Antimony, Barium, Beryllium and Manganese for development of ecological soil screening levels (Ecco-SSL) using folsomia reproduction Benchmark values. ECBC-TR-326, Edgwood chemical biological centre, US Army soldier and biological chemical command. Report no.: ECBC-TR-326.
- Pooles A (2009). MnSO4: Acute Dermal Irritation in the Rabbit. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0121. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2010-01-07.
- Pooles A (2010). MnSO4: ACUTE EYE IRRITATION IN THE RABBIT. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire DE72 2GD, UK. Report no.: 2702/0122. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2010-03-03.
- 86. Rice TM, Clarke RW, Godleski JJ, Al-Mutairi E, Jiang NF, Hauser R and Paulauskis JD (2001). Differential ability of transition metals to induce pulmonary inflammation. Toxicology and Applied Pharmacology, 177: 46-53.
- 87. Roels H, Lauwerys R, Buchet JP, Genet P, Sarhan MJ, Hanotiau I, de Fays M, Bernard A and Stanescu D (1987). Epidemiological survey among workers exposed to manganese: effects on lung, central nervous system, and some biological indices. Am J Ind Med 11:307-327.



88. Salehi F, Krewski D, Mergler D, Normandin L, Kennedy G, Philippe S and Zayed J (2003). Bioaccumulation and locomotor effects of manganese phosphate/sulfate mixture in Sprague-Dawley rats following subchronic (90 days) inhalation exposure. Toxicol Appl Pharmacol 191:264-271.

- 89. Salehi F, Normandin L, Krewski D, Kennedy G, Philippe S & Zayed J (2006). Neuropathology, tremor and electromyogram in rats exposed to manganese phosphate/sulfate mixture. J. Appl. Toxicol., 26: 419-426.
- 90. Sandstrom B, Davidsson L, Eriksson R, Alpsten M and Bogentoft C (1987). Retention of selenium (75Se), Zinc (65Zn) and manganese (54Mn) in humans after intake of a labelled vitamin and mineral supplement. J Trace Elem Electrolytes Health Dis 1:33-38.
- 91. Santucci B, Cannistrachi C, Cristando A and Picardo M (1996). Interactions of sulphates of divalent metals in nickel-sulphate-sensitive patients. Experimental Dermatology. 5:79-83.
- 92. Schneider J, Decamp E, Kosey A, Fritz S, Gonczi H, Syversen T & Guilarte T. (2006). Effects of chronic manganese exposure on cognitive and motor functioning in non-human primates. Brain Research 1118: 222-231.
- 93. SCOEL/SUM/127., (2011); EC recommendation from the scientific committee on occupational exposure limits for manganese and inorganic manganese compounds
- 94. Simini M, Checkai RT, Kuperman RG, Phillips CT, Speicher JA and Barclift DJ (2002). Toxicity assessments of antimony, barium, beryllium and manganese for development of ecological soil screening levels (ECO-SSL) using earthworm (Eisenia fetida) benchmark values. Edgewood chemical biological centre, US army soldier and biological chemical command. Report no.: ECBC-TR-325.
- 95. Singh I (1983). Induction of reverse mutation and mitotic gene conversion by some metal compounds in Saccharomyces cerevisiae. Mutation Research. 117(1-2): 149-152.
- 96. Singh PP and Junnarkar AY (1991). Behavioural and toxic profiles of some essential trace metals salts in mice and rats. Indian Journal of Pharmacology, 23(3): 153-159.
- 97. Standring WJF, Oughton DH & Salbu B (2002). Remobilisation of 109-Cd, 65-Zn and 54-Mn from freshwater-labelled river sediments when mixed with seawater. Environment International 28 (2002) 185-195.
- 98. Streicker MA (2009). In Vivo Micronucleus Assay of Manganese According to OECD 474 Guideline. Testing laboratory: Integrated Laboratory Systems, Inc. 601 Keystone Park Drive, Suite 100, Durham, NC 27713. Report no.: C171-001. Owner Company: Manganese Research Health Project (MHRP). Report date: 2009-09-04.
- 99. Struve MF, McManus BE, Wong BA & Dorman DC (2007). Basal ganglia neurotransmitter concentrations in rhesus monkeys following subchronic manganese sulphate inhalation. American Journal of Industrial Medicine. 50: 772-778. Testing laboratory: CIIT Centers for Health Research, Research Triangle Park, North Carolina 27709.



100. Stubblefield WA, Brinkman SF, Davies PH, Garrison TD, Hockett JR & McIntyre MW (1997). Effects of water hardness on the toxicity of manganese to developing brown trout (Salmo Trutta). Environmental Toxicology and Chemistry. 16(10):2082-2089.

- 101. Tapin D, Kennedy G, Lambert J & Zayed J (2006). Bioaccumulation and locomotor effects of manganese sulfate in Sprague-Dawley rats following subchronic (90 days) inhalation exposure. Toxicology and Applied Pharmacology, 211: 166-174.
- 102. Thompson PW & Bowles A (2009). MnCl2 (Eramet): Reverse Mutation Assay "Ames Test" Using Salmonella Typhimurium and Escherichia Coli. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0035. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2009-09-24.
- 103. Valencia R, Mason JM, Woodruff RC and Zimmering S (1985). Chemical mutagenesis testing in Drosophila. III. Results of 48 coded compounds tested for the National Toxicology Program. Environmental Mutagenesis. 7: 325-348.
- Vryenhoef H (2010). MnO2 (Erachem): Algal Growth Inhibition Test. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0192. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE.
- 105. Warren N (2009a). MnSO4: Determination of Skin Irritation Potential Using the EPISKIN<sup>™</sup> Reconstituted Human Epidermis Model. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0031. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2009-10-28.
- 106. Warren N (2009b). MnSO4: Assessment of Ocular Irritation Potential using the Synthetic Reconstituted Human Corneal Epithelium Model. Testing laboratory: Harlan Laboratories Ltd, Shardlow Business Park, Shardlow, Derbyshire, DE72 2GD, UK. Report no.: 2702-0068. Owner Company: International Manganese Institute, 17 Rue Duphot, 75001 Paris, FRANCE. Report date: 2009-11-06.
- 107. WatlingHR (1983). Comparative Study of the Effects of Metals on the Settlement of Crassostrea gigas. Bull. Environ. Contam. Toxicol. 31, 344-351.
- 108. Webster WS and Valois AA (1987). Reproductive toxicology of manganese in rodents, including exposure during the postnatal period. Neurotoxicology, 8(3): 437-444.
- 109. Witzleben CL, Pitlick P, Bergmeyer J and Benoit R (1968). Acute manganese overload. A new experimental model of intrahepatic cholestasis. Am J Pathol 53:409-422.
- 110. Yasuda H, Uchida S, Muramatsu Y & Yoshida S (1995). Sorption of Manganese, Cobalt, Zinc, Strontium and Cesium. Water, Air and Soil Pollution 83: 85-96, 1995.
- 111. Young A (2009a). Short- Term Chronic Toxicity of Manganese to Duckweed (Lemna minor) Under Static- Renewal Exposure Conditions. Testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas street SW, Albany, OR 97321-9487,



USA. Report no.: 598-5231-002. Owner Company: International Manganese Institute, 17 rue Duphot, 75001 Paris, France. Report date: 2009-03-06.

112. Young A (2009b). Chronic Toxicity of Manganese to the Great Pond Snail, Lymnaea stagnalis. Testing laboratory: Parametrix Environmental Research Laboratory (PERL), 33972 Texas St. SW, Albany, Oregon 97321, USA. Report no.: 598-5231-002. Owner Company: International Manganese Institute, 17 rue Duphot 75001 Paris, France. Report date: 2009-03-11.

- 113. Youngs N (2010). Assessment of the inhibitory effect on the respiration of activated sewage sludge. Testing laboratory: Harlan Laboratories Ltd Shardlow Business Park Shardlow Derbyshire DE72 2GD UK. Report no.: 2702-0189. Owner Company: International Manganese Institute 17 Rue Duphot 75001 Paris FRANCE. Report date: 2010-02-15.
- 114. Zelikoff JT, Atkins N and Rossman TG (1986). Mutagenicity of soluble metal salts using V79/HGPRT Mutation Assay. Environmental Mutagenesis 8(6): 95.

#### Add ANNEX 1: EXPOSURE SCENARIO FOR COMMUNICATION -ask reach@manganese.org

For more information contact: reach@manganese.org